

# Ag-Al-Y (Silver-Aluminum-Yttrium)

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Recently, [2000Gum] determined an isothermal section for this system at 597 °C (870 K), which depicts eight ternary compounds.

(C23, Co<sub>2</sub>Si-type orthorhombic), Y<sub>3</sub>Al<sub>2</sub> (Zr<sub>3</sub>Al<sub>2</sub>-type tetragonal), YAl (*B<sub>f</sub>*, CrB-type orthorhombic), YAl<sub>2</sub> (C15, MgCu<sub>2</sub>-type cubic), αYAl<sub>3</sub> (D0<sub>19</sub>, Ni<sub>3</sub>Sn-type hexagonal), βYAl<sub>3</sub> (stable between 980 and 645 °C, BaPb<sub>3</sub>-type rhombohedral).

## Binary Systems

The Ag-Al phase diagram [Massalski2] depicts three intermediate phases: Ag<sub>3</sub>Al (20.5-29.8 at.% Al; denoted β, body-centered cubic; stable between 778 and ~600 °C), Ag<sub>2</sub>Al (22.9-41.9 at.% Al; denoted δ, close-packed hexagonal), and μ (21-24 at.% Al, stable below 450 °C; A13, βMn-type cubic). The Ag-Y [Massalski2] diagram depicts the following intermediate phases: Ag<sub>51</sub>Y<sub>14</sub> (Ag<sub>51</sub>Gd<sub>14</sub>-type hexagonal), Ag<sub>2</sub>Y (C11<sub>b</sub>, MoSi<sub>2</sub>-type tetragonal), and AgY (*B2*, CsCl-type cubic). The Al-Y phase diagram [2006Liu] depicts the following intermediate phases: Y<sub>2</sub>Al

## Ternary Phases

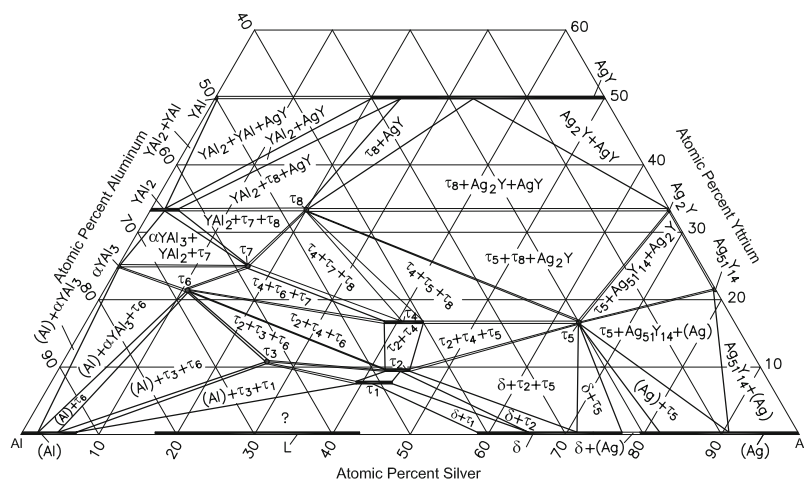
The known ternary compounds of this system are listed in Table 1 from [2000Gum]. The compounds denoted τ<sub>1</sub> to τ<sub>8</sub> (labeled 1 to 8 by [2000Gum]) are present at 597 °C (870 K). The last-listed compound ~YAgAl<sub>3</sub> was not found at 597 °C [2000Gum]. It may be noted that YAgAl<sub>3</sub> and Y<sub>3</sub>Ag<sub>1.5</sub>Al<sub>9.5</sub> (τ<sub>6</sub>) are both derivatives of Al<sub>4</sub>Ba-type of structure. Among the binary phases, AgY and YAl<sub>2</sub> dissolve 30 at.% Al and 7.3 at.% Ag, respectively.

**Table 1** Ag-Al-Y crystal structure and lattice parameter data [2000Gum]

| Phase  | Composition, at.%                     | Pearson symbol | Space group                   | Prototype   | Lattice parameter, nm  |
|--|---------------------------------------|----------------|-------------------------------|---|--|
| YAg <sub>5.1-5.7</sub> Al <sub>6.9-6.3</sub> (τ <sub>1</sub> )                 | 39.2-43.8 Ag<br>53.1-48.5 Al<br>7.7 Y | <i>tI26</i>    | <i>I4/mmm</i>                 | ThMn <sub>12</sub>                                | <i>a</i> = 0.91303*<br><i>c</i> = 0.54646                      |
| Y <sub>1.8</sub> Ag <sub>7.9-8.5</sub> Al <sub>9.1-8.5</sub> (τ <sub>2</sub> ) | 42.0-45.2 Ag<br>48.4-45.2 Al<br>9.6Y  | <i>hP38</i>    | <i>P6<sub>3</sub>/mmc</i>     | Th <sub>2</sub> Ni <sub>17</sub>                  | <i>a</i> = 0.92852*<br><i>c</i> = 0.90702                      |
| Y <sub>8</sub> Ag <sub>19.4</sub> Al <sub>46.6</sub> (τ <sub>3</sub> )         | 26.2 Ag<br>63.0 Al<br>10.8 Y          | <i>tI?</i>     | <i>I4/mmm</i>                 | Yb <sub>8</sub> Cu <sub>17</sub> Al <sub>49</sub> | <i>a</i> = 0.88278<br><i>c</i> = 1.66993                       |
| YAg <sub>2.3-2.6</sub> Al <sub>2.7-2.4</sub> (τ <sub>4</sub> )                 | 38.3-43.3 Ag<br>45-40 Al<br>16.7 Y    | <i>hP?</i>     | <i>P6<sub>3</sub>/mmc</i>     | DyAg <sub>2.4</sub> Al <sub>2.6</sub>             | <i>a</i> = 0.91523*<br><i>c</i> = 0.94167                      |
| ~YAg <sub>3.8</sub> Al <sub>1.2</sub> (τ <sub>5</sub> )                        | 63.3 Ag<br>20 Al<br>16.7 Y            | Hex.           | ...                           | ...   | <i>a</i> = 0.5362<br><i>c</i> = 0.9198                         |
| Y <sub>3</sub> Ag <sub>1.5</sub> Al <sub>9.5</sub> (τ <sub>6</sub> )           | 10.7 Ag<br>67.85 Al<br>21.4 Y         | <i>oI28</i>    | <i>Immm</i>                   | αLa <sub>3</sub> Al <sub>11</sub>                 | <i>a</i> = 0.42827<br><i>b</i> = 1.26026<br><i>c</i> = 1.00312 |
| Y <sub>3</sub> Ag <sub>2</sub> Al <sub>7</sub> (τ <sub>7</sub> )               | 16.7 Ag<br>58.3 Al<br>25 Y            | <i>hR12</i>    | <i>R<math>\bar{3}m</math></i> | Ca <sub>3</sub> Cu <sub>2</sub> Al <sub>7</sub>   | <i>a</i> = 0.55422<br><i>c</i> = 2.6259                        |
| YAg <sub>0.6</sub> Al <sub>1.4</sub> (τ <sub>8</sub> )                         | 20 Ag<br>46.7 Al<br>33.3 Y            | <i>oI12</i>    | <i>Imma</i>                   | CeCu <sub>2</sub>                                 | <i>a</i> = 0.45399<br><i>b</i> = 0.71551<br><i>c</i> = 0.78712 |
| YAgAl <sub>3</sub>   | 20 Ag<br>60 Al<br>20 Y                | <i>oI?</i>     | <i>Immm</i>                   | CeNi <sub>2</sub> Sb <sub>2</sub>                 | <i>a</i> = 0.4295<br><i>b</i> = 0.4184<br><i>c</i> = 1.0039    |

\* Lattice parameters are for YAg<sub>5.2</sub>Al<sub>6.8</sub>, Y<sub>1.8</sub>Ag<sub>8.3</sub>Al<sub>8.7</sub>, and YAg<sub>2.3</sub>Al<sub>2.7</sub>, respectively

## Section II: Phase Diagram Evaluations



**Fig. 1** Ag-Al-Y isothermal section at 597 °C (870 K) [2000Gum]

### Isothermal Section

With starting metals of 99.95% Ag, 99.99% Al, and 99.5% Y, [2000Gum] arc-melted under Ar atm 69 alloys with Y content up to 50 at.%. The alloys were annealed at 597 °C (870 K) for ~1000 h and quenched in water. The phase equilibria were studied mainly with x-ray powder diffraction. The isothermal section at 597 °C is redrawn in Fig. 1, to agree with the accepted binary data. The

compositions of the ternary phases shown in Fig. 1 are from the listed values of [2000Gum].

### References

- 2000Gum:** T.M. Gumenyuk, Yu.B. Kuzma, and B.M. Stelmakhovych, The Y-Ag-Al System, *J. Alloys Compd.*, 2000, **299**, p 213-216  
**2006Liu:** S. Liu, Y. Du, H. Xu, C. He, and J.C. Schuster, Experimental Investigation of the Al-Y Phase Diagram, *J. Alloys Compd.*, 2006, **414**, p 60-65